



Factorial experimental design for optimizing the removal of lead ions from aqueous solutions by cation exchange resin

Duygu Kavak^{a,*}, Melek Demir^a, Burak Başsayel^a, A. Sermet Anagün^b

^aFaculty of Engineering and Architecture, Department of Chemical Engineering, Eskişehir Osmangazi University, Meselik, Eskişehir 26480, Turkey

Tel. +90 222 2393750/3645; Fax: +90 222 2393613; email: dbayar@ogu.edu.tr

^bFaculty of Engineering and Architecture, Department of Industrial Engineering, Eskişehir Osmangazi University, Meselik, Eskişehir 26480, Turkey

Received 29 February 2012; Accepted 18 July 2012

ABSTRACT

Full factorial design of experiments were used to screen the factors affecting the lead removal efficiency using cation exchange resin. The obtained linear model was statistically tested using analysis of variance (ANOVA), Student's *t*-test, lack of fit test, and test of curvature. The percentage removal of lead was examined by varying experimental conditions with center points. The factors and levels used during the experiments were; initial pH (3.5, 4.5 and 5.5), temperature (25, 35 and 45 °C), initial lead concentration (20, 60, and 100 mg/L) and resin dosage (0.02, 0.26, and 0.5 g). A steepest ascent based optimization procedure was implemented to seek better conditions in terms of maximizing the removal of lead(II) ions. The results showed that approximately 99% removal of Pb(II) was obtained when initial pH, temperature, initial lead concentration, and resin dosage are roughly set to 5, 34, 32, and 0.74, respectively.

Keywords: Lead; Heavy metal removal; Cation exchange resin; Factorial design; Optimization; Steepest ascent

*Corresponding author.