



## Simulation of a full-scale wastewater treatment plant performance at various temperatures using Extended Activated Sludge Model No.1

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### ABSTRACT

In this paper, a simulation model was adapted for the 6th of October wastewater treatment plant (WWTP) located in Cairo, Egypt. The study aims to investigate the effect of variable temperature on treated effluent wastewater with regards to various operational parameters. The temperature in the arid region where the plant is located varies significantly between summer and winter. The simulation model was created using the Runge–Kutta 4th order numerical technique in the MATLAB platform and applies Extended Activated Sludge Model No. 1 (ASM1) equations for WWTP modeling. This model reflected the equations of ASM1 with modified Monod kinetics for the dissociation of soluble biodegradable organic substrates into unionized organic substrates to be utilized by aerobic heterotrophs and autotrophs. Wastewater characterization was determined with a detailed sampling campaign for the WWTP influent, primary settled, and effluent discharges. Model calibration and sensitivity analyses of kinetic parameters were conducted for model validation. A comprehensive study was performed to examine the effect of various operating parameters on the removal of chemical oxygen demand (COD) at different temperatures using the calibrated model. The operational conditions studied are dissolved oxygen (DO), hydraulic retention time (HRT), and mixed liquor volatile suspended solids (MLVSS). The modified model indicates that COD removal efficiency reaches about 65% for MLVSS of 2,500 mg/L at 15°C, whereas the COD removal efficiency is increased by about 15% as the measured temperature increases from 15°C to 35°C at a DO value of 2 mg/L and HRT of 6 h. Furthermore, the simulation results imply that COD removal can be significantly enhanced by increasing both MLVSS and DO to 4,000 and 3 mg/L, respectively.

*Keywords:* COD; Extended Activated Sludge Model No. 1; Operating parameters; Temperature; WWTP

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### 1. Introduction

Wastewater treatment is essential for minimizing disease transmission resulting from bacterial and viral reproduction as well as protecting water resources and allowing their use for different purposes. In particular, biological wastewater treatment processes depend on accelerating

natural degradation processes for organic contaminants by either aerobic or anaerobic microorganisms under controlled conditions. Among those, the activated sludge process, the integrated fixed-film activated sludge (IFAS) process, anaerobic digestion, and a two-step anaerobic–anoxic and oxic (A<sup>2</sup>O) process are commonly applied in Egypt for removal of both organic matter and nutrients [1,2]. Low-cost

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wastewater treatment methods are also used in Egypt. These methods include the oxidation-ponds processes that require a hot climate which prevails most of the year in Egypt. The oxidation-ponds processes have high land area requirements, which are addressed by constructing the ponds in the desert where vast amounts of land are available. Some examples of such low-cost treatment methods are the El Sadat treatment plant, which uses anaerobic, facultative, and maturation ponds [3] and the Al Ismailia plant consisting of an aerated lagoon, a facultative aerated lagoon, and polishing ponds [4].

After many decades of applying treatment methods, a computerized method to simulate the performance of a full-scale wastewater treatment plant (WWTP) was needed. Mathematical modeling can provide plant operators with highly accurate, predicted values of a plant's performance that would be useful in making decisions about the different operating parameters. In the early eighties, the International Association of Water Quality (IAWQ) formed a task group to develop the simplest mathematical model for the prediction of activated sludge system performance. In 1987, the activated sludge system model was created and named Activated Sludge Model No. 1 (ASM1). Since then, activated sludge models (ASMs) have evolved enormously. The most well-known ASMs are ASM1, ASM2, ASM2d, and ASM3, all of which are used often in simulation studies for municipal and industrial WWTPs [5–7]. An ASM is a representation of microbial growth and substrate utilization within an activated sludge system through a dynamic mathematical expression [8,9].

ASMs are incorporated through various simulation platforms such as Simba, GPS-X, and BioWin. ASM equations can be computer-generated with the MATLAB programming platform utilized in this study. Simulating the performance of a full-scale WWTP is done either to enhance the quality of the plant's effluent or to study the relationship between parameters that affect its performance. Enhancing the quality of effluent at the 6th of October WWTP is of major importance due to the environmental policies adopted by the Egyptian government. The significant variation in temperature in the arid region of the plant's location affects the effluent quality of the treated wastewater at this plant. This study aims at studying the effect of temperature on treated effluent wastewater with regards to various operational parameters. MATLAB version R2016b (9.10.441655) software is utilized to create equations that simulate the actual performance of the 6th of October WWTP using extended ASM1 [10,11]. The conventional ASM1 only considers the hydrolysis of particulate (slowly) biodegradable organic substrates into soluble (readily) biodegradable compounds to be utilized by aerobic heterotrophs and autotrophs. The role of pH in the dissociation of soluble organic substrates is ignored in this version of the model. In this paper, the extended model simulates the removal of chemical oxygen demand (COD) by considering the hydrolysis of soluble organic substrates such as acetic acid into unionized substrates utilized by bacteria. Moreover, the efficiency of the WWTP with regard to the total COD percentage of removal is studied relative to the following operational parameters: hydraulic retention time (HRT), dissolved oxygen (DO), and mixed liquor volatile suspended solids (MLVSS).

Previous studies were conducted to investigate the effect of HRT, DO, and MLVSS on pilot-scaled studies. None of these studies was conducted on a full-scale WWTP, nor was the effect of these operational parameters on the removal efficiency compared for the same plant at different temperatures. One study was conducted on a pilot reactor to investigate the effect of HRT on removal efficiency in Alexandria, Egypt [12]. The removal efficiency of the plant was enhanced by decreasing the HRT; however, the paper stated that this enhancement was due to the accumulation of biomass that occurred during the operation of the plant at higher HRTs. Another study was conducted on a pilot reactor to investigate the influence of HRT on the removal efficiency of COD,  $BOD_5$ , total suspended solids (TSS), and  $NH_4-N$  [2,13]. The results showed that the removal efficiency increased by raising the HRT. A pilot-scaled reactor was also used to investigate the effect of MLVSS and HRT on the removal efficiency of a WWTP [14]. The results showed that the performance of the pilot reactor was enhanced by increasing both operational parameters. The aforementioned studies were all conducted on experimental-scale reactors. The aim of our study is to predict the performance of a full-scale WWTP using an extended ASM1 in conjunction with the data collected from the plant under various operational parameters at various temperature segments.

## 2. Materials and methods

Although there are many modified ASMs, ASM1 remains one of the most reliable simulation models and is the most widely used worldwide [11,15,16]. In ASM1, carbon material is categorized as either biodegradable, non-biodegradable COD, or active biomass, as shown in Fig. 1.

### 2.1. Description of the 6th of October WWTP process

The WWTP under study is located in 6th of October, a city in Giza Governorate, Egypt. It began operating in 1988 and serves 13 residential districts, ten compounds,

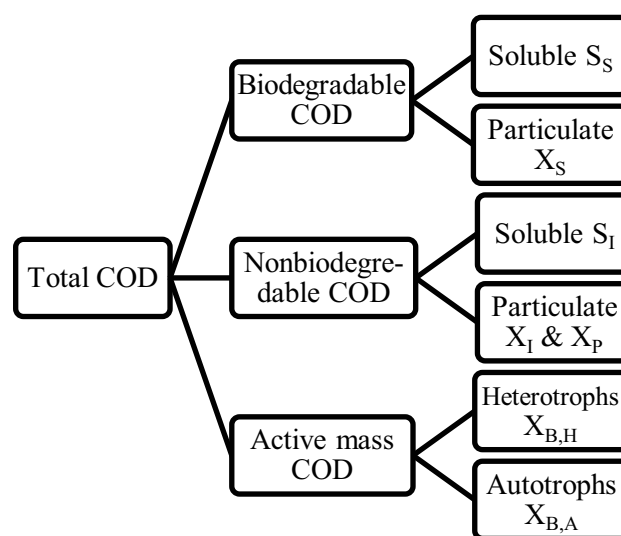


Fig. 1. Segregation of total COD as per ASM1 equations.

and an industrial zone. The plant is operated by applying the conventional activated sludge treatment method. The current design of the plant consists of three modules, each with a capacity of 50,000 m<sup>3</sup>/d. As such, the current capacity of the plant is 150,000 m<sup>3</sup>/d, whereas the future design capacity is expected to reach 600,000 m<sup>3</sup>/d. The total current influent to the plant is 120,000 m<sup>3</sup>/d. Note that 65% of the influent is domestic sewage and the remaining 35% is industrial wastewater, which includes different types of waste (i.e., leather, textile, food, chemical, and pharmaceutical industries). Each module includes primary sedimentation tanks (PSTs), aeration tanks, final sedimentation tanks (FST), chlorination tanks, and rapid sand filters. Fig. 2 shows a schematic diagram of the WWTP.

The historical data for this plant shows that the total COD removal efficiency following tertiary treatment reaches 90%. The treated sewage effluent is used in the irrigation of boundary trees. This study relied on historical readings acquired from the plant's records that were used to analyze its wastewater quality. Wastewater characterization was identified to generate the model calibration, validation, and sensitivity analysis. Data were collected from 425 archive readings of the WWTP during the period from April 1, 2013, to May 31, 2014. Table 1 illustrates the average values of the different wastewater characteristics for influent sewage, primary settled wastewater, and secondary settled wastewater effluent based on the actual readings of the WWTP. Table 2 illustrates the wastewater characteristics inside the aeration tanks, where an average DO value of 1.55 mg/L and a pH value of 8.3 were measured. These values are also based on the actual readings of the WWTP. The average pH value of the influent wastewater to the plant is relatively

high due to the industrial wastewater influent that may reach up to 35% of the plant's total influent [17].

Samples were collected from September 20 to 25, 2014, for more relevant values. The collected samples underwent testing at the Western 6th of October WWTP laboratory and the National Research Center Laboratory in Cairo, Egypt. Samples were collected at two points: influent to the activated sludge tanks from the PST and effluent of the FST. The following tests were included for both the influent and effluent samples: pH, TSS, volatile suspended solids (VSS), TS, total volatile solids (TVS), total dissolved solids (TDS), BOD<sub>5</sub>, COD, filtered COD, DO, total Kjeldahl nitrogen (TKN), NO<sub>2</sub>-N, NO<sub>3</sub>-N, and NH<sub>3</sub>-N. In addition, total alkalinity, total phosphorus, and ortho-phosphorus tests were performed for the influent samples only. Tables 3 and 4 illustrate the values of the sampling program conducted at the two sampling points. The HRT of the activated sludge system under study was calculated using the historical data

Table 2  
Wastewater characteristics inside the aeration tank

Parameter	Inside the Aeration tank
MLSS (mg/L)	4,049 ± 641
MLVSS (mg/L)	2,791 ± 673
F/M ratio	0.16 ± 0.13
SRT (days)	11.74 ± 7.48
RAS VSS (mg/L)	5,895 ± 1,242
RAS TSS (mg/L)	7,448 ± 1,876

Based on actual measurements in the WWTP.

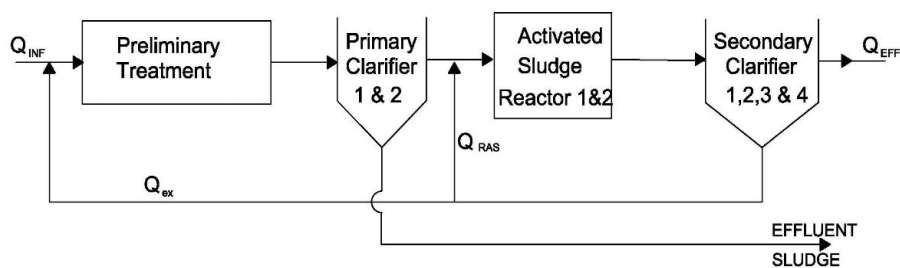


Fig. 2. Schematic diagram of the WWTP under study.

Table 1  
Wastewater characteristics of the 6th of October WWTP

Parameter	Influent to the WWTP	Primary settled	Secondary settled
BOD <sub>5</sub> (mg/L)	491 ± 91	269 ± 50	36.33 ± 5
COD (mg/L)	1,464 ± 256	465 ± 80	96.85 ± 17
TSS (mg/L)	458 ± 102	181.5 ± 38	37.7 ± 6
VSS (mg/L)	406 ± 84	134 ± 36	32.55 ± 25
TS (mg/L)	1,648 ± 469	1,165 ± 344	592 ± 142
TVS (mg/L)	1,105 ± 310	720 ± 358	234 ± 134
TDS (mg/L)	434 ± 171	437 ± 101	420 ± 92

Based on actual measurements in the WWTP.

Table 3  
Measured parameters of the PST effluent

Date	20-09-2014	21-09-2014	22-09-2014	23-09-2014	24-09-2014	25-09-2014
Parameter						
T (°C)	28.6	29.9	29.3	28.3	29.3	28.2
Flow (m <sup>3</sup> /d)	49,500	49,000	49,750	49,750	49,250	49,250
pH	8.7	9.3	8.9	8.9	8.7	9
TSS (g/m <sup>3</sup> )	352	408	293	353	312	373
VSS (g/m <sup>3</sup> )	281	343	212	305	281	301
TS (g/m <sup>3</sup> )	680	820	653	715	742	816
TVS (g/m <sup>3</sup> )	420	680	401	663	671	786
TDS (g/m <sup>3</sup> )	228	305	360	416	395	422
BOD <sub>5</sub> (g/m <sup>3</sup> )	225	217	174	215	190	220
COD (g/m <sup>3</sup> )	548	560	498	449	397	447
Filtered COD (g/m <sup>3</sup> )	260	275	238	219	195	214
DO (g/m <sup>3</sup> )	0.3	0	0.2	0	0.2	0.1
TKN (g/m <sup>3</sup> )	32.3	33.9	35.6	34.1	34.4	34.8
NO <sub>2</sub> -N (g/m <sup>3</sup> )	0.03	0.024	0.053	0.082	0.105	0.008
NO <sub>3</sub> -N (g/m <sup>3</sup> )	0.039	0.066	0.069	0.01	0.24	0.184
NH <sub>3</sub> -N (g/m <sup>3</sup> )	0.21	1.6	0.06	1.8	0.039	1.02
Total alkalinity	285	440	270	160	300	285
Total P	2.34	1.24	3.24	1.76	3.88	2
Ortho P	1.45	0	1.97	0	2.3	1.4

Based on the sampling program.

Table 4  
Measured parameters of the FST effluent

Date	20-09-2014	21-09-2014	22-09-2014	23-09-2014	24-09-2014	25-09-2014
Parameter						
pH	8.8	8.9	8.7	8.7	8.7	8.8
TSS (g/m <sup>3</sup> )	35	47	36	28	25	31
VSS (g/m <sup>3</sup> )	28	41	31	21	18	22
TS (g/m <sup>3</sup> )	320	452	362	411	425	186
TVS (g/m <sup>3</sup> )	300	425	341	389	376	402
TDS (g/m <sup>3</sup> )	260	308	300	308	335	395
BOD <sub>5</sub> (g/m <sup>3</sup> )	32	42	31	24	31	42
COD (g/m <sup>3</sup> )	75	92	80	73	75	93
Filtered COD (g/m <sup>3</sup> )	30	37	32	29	30	37
NO <sub>2</sub> -N (g/m <sup>3</sup> )	0.245	0.062	0.3	0.225	0.37	0.21
NO <sub>3</sub> -N (g/m <sup>3</sup> )	0.84	0.07	1.5	0.72	0.48	0.685
NH <sub>3</sub> -N (g/m <sup>3</sup> )	0.13	0.15	0.1	0.01	0.05	0.03

Based on the sampling program.

provided and found at 6.14–6.47 h. The physical parameters of pH and temperature have a major effect on the wastewater treatment process. Temperature varies significantly due to normal seasonal variations in the arid location of the plant. Thus, it was important to analyze the temperature values (less than 15°C, 15°C–25°C, 25°C–35°C, and greater than 35°C) throughout the study period. Furthermore, the

6th of October WWTP receives both domestic and industrial wastewater, resulting in high pH values at a range of 7.2–10. The typical values of kinetic and stoichiometric parameters were considered at neutral pH values.

Both the actual readings and the samplings measurements program showed a significant percentage of removal of COD in the PST. This can be attributed to the characteristics

of the plant's influent that includes industrial wastewater of high amount of particulate matter that sediments in the primary sedimentation tanks.

## 2.2. Model approach and calibration

Several computer programs can apply the ASM equations in WWTP simulations, including WEST, STOAT, and Simba. Another programming platform that can be used for simulation processes is MATLAB, which is designed specifically for engineers and scientists. MATLAB utilizes a matrix-based language that allows the most natural expression of computational mathematics. The Runge–Kutta fourth order numerical technique was applied in the extended model using MATLAB software. This model reflected the equations of ASM1 with modified Monod kinetics for the hydrolysis of soluble biodegradable organic substrates [10,11]. The extended ASM1 indicates that soluble organic substrates are dissociated into unionized organic substrates to be utilized by aerobic heterotrophs and autotrophs that maintain a constant pH. Based on El-Seddik [10] and Al Madany et al. [11], the biomass and substrate mass balance equations of extended ASM1 are expressed as follows:

$$V \frac{dC_{X(H,A)}}{dt} = F_{in} (C_{Xi(H,A)} - C_{X(H,A)}) + (\mu_{H,A} - K_{d(H,A)}) C_{X(H,A)} V \quad (1)$$

$$V \frac{dC_{S(H,N)}}{dt} = F_{in} (C_{Si(H,N)} - C_{S(H,N)}) - \left( \frac{\mu_{H,A}}{Y_{H,A}} - K_{d(H,A)} \right) C_{X(H,A)} V \quad (2)$$

where the specific growth rate of heterotrophic and autotrophic bacteria can be expressed as follows [10,11]:

$$\mu_{H,A} = \mu_{\max(H,A)} \frac{C_{HS(H,N)}}{K_{S,N} + C_{HS(H,N)}} \times \frac{C_{O_2}}{K_{O_2(H,A)} + C_{O_2}}, \quad (3)$$

$$C_{HS(H,N)} = \frac{10^{-\text{pH}} C_{S(H,N)}}{K_A + 10^{-\text{pH}}}$$

Model calibration, validation, and a sensitivity analysis were generated to ensure the adequacy of the created model relative to the results of the plant. The MATLAB model simulates the kinetic and biological reaction occurring inside the biological treatment process. The influent value of any parameter addressed in the model represents the value of the primary settled wastewater effluent characteristics. The effluent of any parameter is the effluent from the FST. Two types of parameters are involved in the extended ASM1 model, stoichiometric and kinetic. Stoichiometric parameters are temperature-independent parameters that help to calculate the relationship between the relative quantities of substances taking part in a reaction. Kinetic parameters are temperature-dependent parameters [17]. The kinetic parameters of extended ASM1 include  $\mu_{\max(H)}$ ,  $\mu_{\max(A)}$ ,  $K_{d(H)}$ ,  $K_{d(A)}$ ,  $K_{S'}$ ,  $K_{N'}$ ,  $K_{O_2(H)}$ , and  $K_{A'}$  while the stoichiometric parameters considered in the extended model are  $Y_H$  and  $Y_A$ . The dissociation coefficient ( $K_A$ ) can be attributed to Atkins [18] and El-Seddik et al. [19] as reported by Al Madany et al. [11].

The calibration process aims to ensure the parameters used in the MATLAB model are appropriate for this specific WWTP. Many protocols are developed and used for efficient process simulations (e.g., STOWA, WERF, and BIOMATH). The protocol adopted herein is STOWA, which determines wastewater characterization as a ratio to total COD as follows: 3%, 45%, 30%, and 22% for  $S_H$ ,  $S_S$ ,  $X_H$ , and  $X_S$ , respectively, as reported by Sin and Vanrolleghem [20] and Elawwad et al. [7]. The model was calibrated using the typical values of kinetic parameters as well as the stoichiometric parameters indicated in Table 5. Several iterations were performed using the MATLAB model to approach the closest values for the total effluent COD concentration. One of the most important parameters considered was the maximum specific growth rate of both heterotrophic and autotrophic biomass ( $\mu_{\max(H,A)}$ ), which is the relation between substrate degradation capacity and biomass growth under specific treatment conditions such as BOD, pH, nutrients, and alkalinity. The Hoff–Arrhenius equation (Eq. (4)) shows that the  $\mu_{\max(H,A)}$  depends on the temperature ( $T$ ). For an activated sludge system, the temperature-activity coefficient ( $\theta$ ) is taken as 1.04 [21]. Eq. (4) helps to deduce the values of kinetic parameters at a specific temperature [7,20].

$$\mu_{\max(T)} = \mu_{\max(20)} \theta^{(T-20)} \quad (4)$$

Heterotrophic biomass is an extremely important parameter to measure and include in the model. The value of the heterotrophic active biomass concentration due to mixed liquor ( $X_{BH}$ ) should be calculated for adequate calibration and validation. Based on experimental measurements, Lee et al. [22] addressed an equation to calculate the heterotrophic biomass concentration in an aeration tank. Eq. (5) was utilized to predict the value of  $X_{BH}$  as follows:

$$X_{BH} = \frac{e^{(Y-\text{intercept})}}{(f_{CV} + 4.57 f_N)(1 - f_E)(-\text{slope})} \quad (5)$$

where  $f_E$  is the fraction of heterotrophic active biomass that is endogenous residue = 0.2;  $f_{CV}$  is the COD to VSS ratio of the mixed liquor organic suspended solids (mgCOD/mgVSS);  $f_N$  is the TKN to VSS ratio of the mixed liquor organic suspended solids (mgTKN/mgVSS);  $Y$ -intercept = 2.5 (for oxygen uptake rate (OUR) = 12 mg O/L/h); slope =  $-3/240$  (with respect to the time period).

The appropriate kinetic and stoichiometric parameters, which reflect the proper performance of this specific WWTP, could be deduced using the calculated average values of  $X_{BH}$  and the maximum specific growth rates of heterotrophs and autotrophs. A set of runs were conducted to validate the appropriate values of the kinetic and stoichiometric parameters for this plant. Table 5 shows the calibrated values of different parameters.

## 2.3. Effect of temperature on various operating conditions

Variation in temperature affects the maximum specific growth rates of both heterotrophs and autotrophs. The values of  $\mu_{\max(H,A)}$  is taken as per the typical literature values

Table 5  
Calibrated values of kinetic and stoichiometric parameters for the WWTP under study

Model parameters	IAWQ model parameter definition	Literature values		Calibrated values
		20°C	10°C	
$\mu_{\max(H)}$	Maximum heterotrophic specific rate of growth, $d^{-1}$	6.4	3	5.54
$\mu_{\max(A)}$	Maximum autotrophic specific rate of growth, $d^{-1}$	0.8	0.3	0.73
$K_{d(H)}$	Decay rate for heterotrophs, $d^{-1}$	0.6	0.05	0.5
$K_{d(A)}$	Decay rate for autotrophs, $d^{-1}$	0.2	0.05	0.2
$K_S$	Monod half-saturation coefficient for heterotrophs, $g\ COD\ m^{-3}$	60	20	30
$K_{O_2(H)}$	Oxygen half-saturation coefficient for heterotrophs, $g\ O_2\ m^{-3}$	0.2	0.1	0.2
$K_A$	Dissociation coefficient	$\leq 0.0001$		0.000001
$K_N$	Monod half-saturation constant for autotrophs, $g\ NH_3-N/m^3$	1.0	1.0	1.0
$Y_H$	Yield coefficient for heterotrophs, $g\ cell\ COD\ formed/g\ COD\ oxidized$	0.67	0.67	0.69
$Y_A$	Yield coefficient for autotrophs, $g\ cell\ COD\ formed/g\ N\ oxidized$	0.14	0.14	0.14

at 20°C. Using Eq. (4), the maximum specific growth rate can be deduced for several temperature intervals. Tables 5 and 6 show the typical values of kinetic parameters at 20°C along with the deduced values of the maximum specific growth rates for both heterotrophs and autotrophs according to the temperature range. Operating temperatures were categorized into five ranges (less than 15°C, 15°C–20°C, 20°C–25°C, 25°C–35°C, and greater than 35°C). These temperature ranges are actual values from the plant's records as the temperature varies significantly during the year. The maximum heterotrophic and autotrophic biomass concentrations were deduced for each run based on these temperature ranges. The operating parameters studied were MLVSS, DO, and HRT. Each parameter was studied in a separate set of runs. Calibrated stoichiometric and kinetic parameters were used during each set of runs to match the performance of the actual full-scale WWTP under study.

### 3. Results and discussion

#### 3.1. Extended model validation and sensitivity analysis

A set of runs was created for the 400 historical readings of the WWTP to ensure appropriate total effluent COD results after applying the calibrated parameters. Simulated effluent soluble COD is converted into total effluent COD according to the STOWA protocol. Fig. 3 shows the actual total effluent COD from primary and final settling tanks as

per the historical readings and the simulated total effluent COD from the aeration tank. The results show that the simulated total effluent COD agrees with that of the measured outcomes from the WWTP, indicating the robustness of the extended ASM1.

Fig. 4 shows the simulation results of soluble COD removal and heterotrophic biomass concentration in the aeration tank for six different actual readings of the WWTP. Note that the simulation below is done using the calibrated kinetic and stoichiometric parameters, whereas the measured operating conditions for these runs are mentioned in Table 7. The simulation results indicate that effluent soluble COD reaches about 50 mg/L for influent soluble COD to the aeration tank fluctuating between 170 and 280 mg/L as shown in Fig. 4a. Also, Fig. 4b shows an increase in the heterotrophic biomass concentration in the aeration tank from 4,000 to 8,000 mg/L that can be attributed to the variation of MLVSS in the aeration tank. However, the steady-state results of effluent soluble COD approach 0.05 kg/m<sup>3</sup> due to the different operational conditions mentioned in Table 7.

Sensitivity analysis should be incorporated into the calibration protocol to minimize the amount of effort needed to optimize the calibration procedure, according to Sin and Vanrolleghem [20]. Sensitivity analysis is performed in the steady-state calibration stage and used to determine which parameters of the model have the greatest influence on the plant's performance and, therefore, the significant need for calibration. Accordingly, the sensitivity analysis was

Table 6  
Maximum specific growth rate of bacteria at different temperature intervals

T (°C)	Average temp. (°C)	Autotrophs		Heterotrophs	
		$\mu_{\max(20)}$ ( $d^{-1}$ )	$\mu_{\max(T)}$ ( $d^{-1}$ )	$\mu_{\max(20)}$ ( $d^{-1}$ )	$\mu_{\max(T)}$ ( $d^{-1}$ )
<15	13.6	0.8	0.622	6.0	4.668
15–20	17.98	0.8	0.739	6.0	5.544
20–25	22.50	0.8	0.882	6.0	6.618
25–30	27.49	0.8	1.073	6.0	8.048
30–35	31.38	0.8	1.250	6.0	9.377

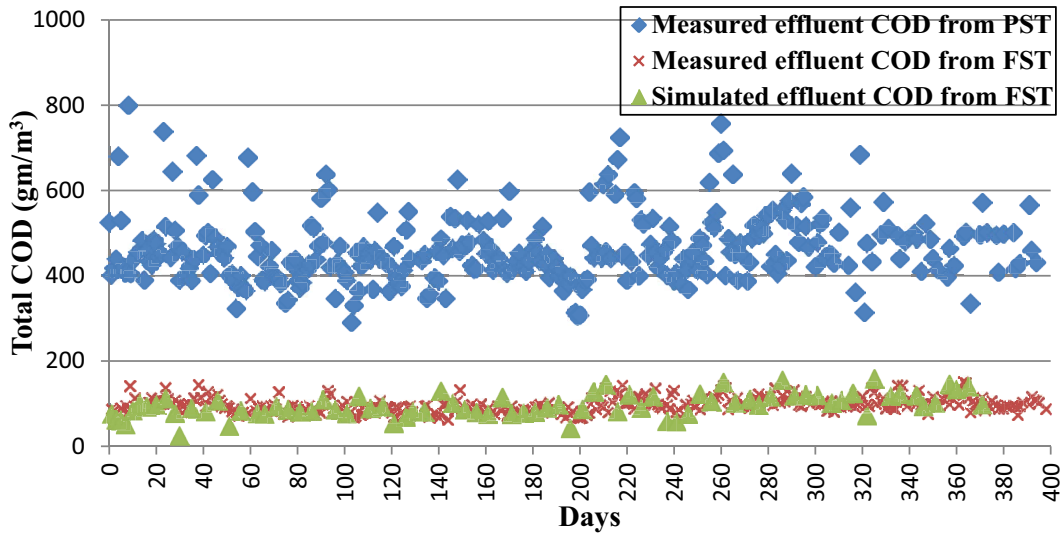


Fig. 3. Measured influent and effluent COD values and simulated effluent COD values of the aeration tank during the WWTP study period.

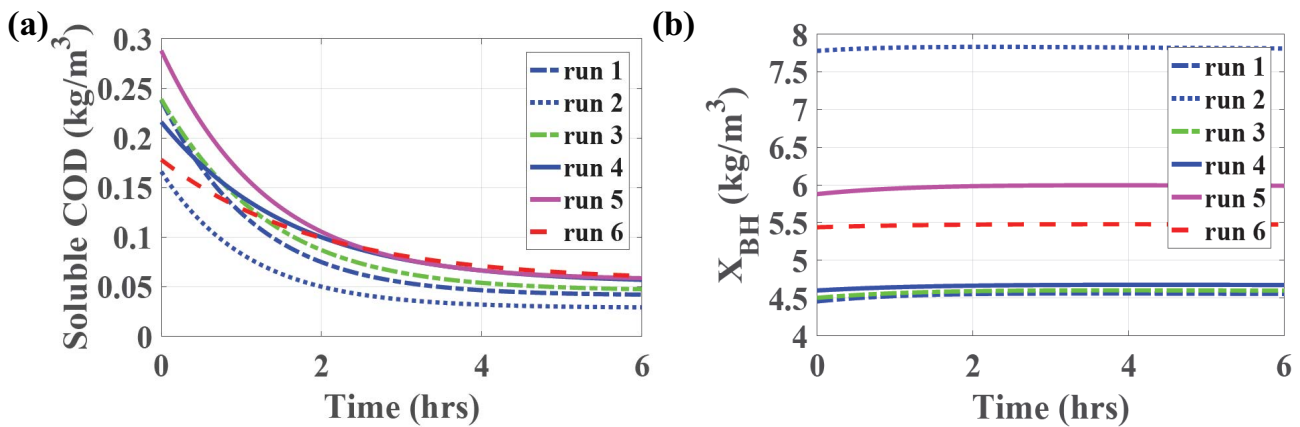


Fig. 4. MATLAB simulation results for (a) soluble COD and (b) heterotrophic biomass concentration of randomly chosen actual readings.

Table 7  
Measured operating conditions for different runs used in the extended ASM1

	pH	T (°C)	Total COD influent (g/m <sup>3</sup> )	Soluble COD influent (g/m <sup>3</sup> )	MLVSS (g/m <sup>3</sup> )	DO (mg/L)	Simulated COD effluent (g/m <sup>3</sup> )	Total COD effluent (g/m <sup>3</sup> )
Run 1	9.3	22.6	529.92	238.46	2,510	1.8	37.8	84
Run 2	7.6	32	368.32	165.74	3,347	1.2	35.1	78
Run 3	7.4	12.8	531.2	239.04	2,540	1.4	48.6	108
Run 4	8.2	18.4	481.92	216.86	2,410	1.6	58.95	131
Run 5	8.2	19.6	640	288	3,842	1	58.95	131
Run 6	9.4	26.3	397	178.65	2,470	1.3	44.1	98

performed on one variable at a time. During this process, one variable is changed, and the others remain constant at the literature review average values. Fig. 5 shows the results of the sensitivity analysis; the horizontal axis represents the dimensionless input of the variable parameter,

and the vertical alignment represents the dimensionless value for the parameter under study. The dimensionless value represents the value of the parameter per run divided by the average value or the proposed literature value at a convenient temperature. The output value represents the

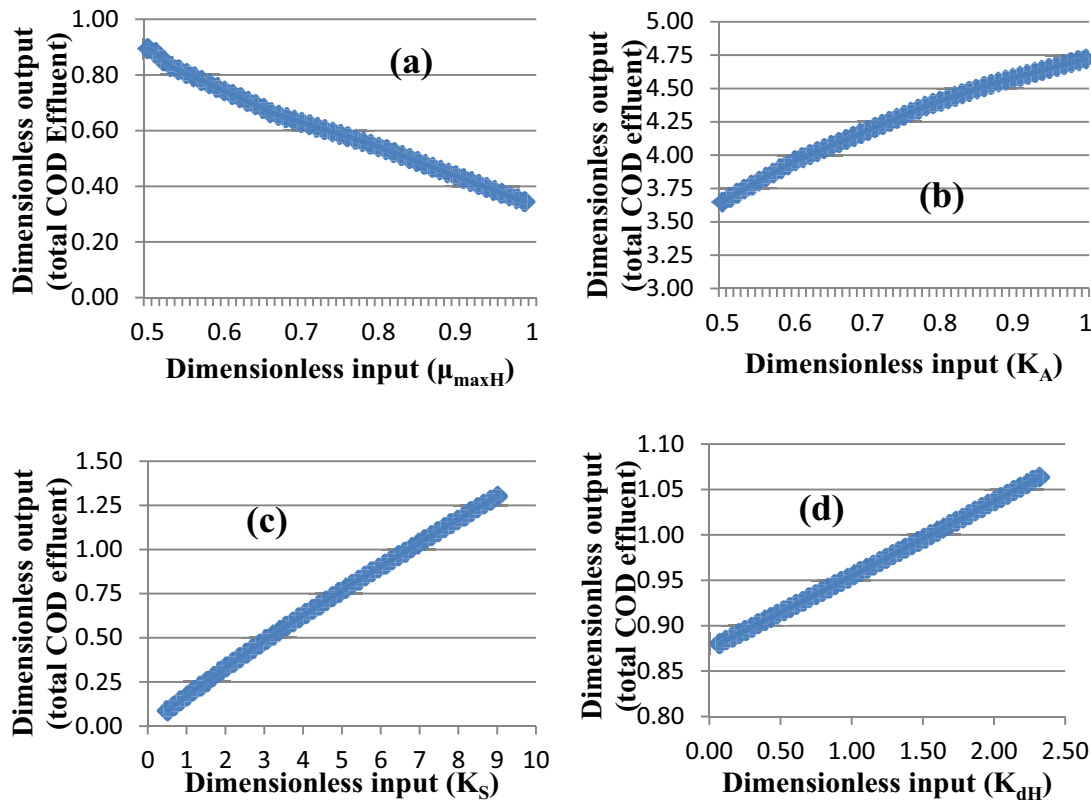


Fig. 5. Sensitivity analysis for (a) maximum heterotrophic specific rate of growth ( $\mu_{\max(H)}$ ), (b) dissociation coefficient ( $K_A$ ), (c) Monod half-saturation coefficient for heterotrophs ( $K_S$ ), and (d) decay rate for heterotrophs ( $K_{d(H)}$ ).

total effluent COD concentration. The sensitivity analysis revealed that  $K_S$  and  $\mu_{\max(H)}$  are the parameters with the greatest influence on total effluent COD, whereas  $K_A$  and  $K_{d(H)}$  have less impact on effluent COD [23].

### 3.2. Effect of temperature on WWTP performance

Utilizing the parameters calculated above, as well as the calibrated kinetic and stoichiometric parameters mentioned in Tables 5 and 6, an assessment for the effect of the various operating parameters on COD removal is performed at different temperatures using the extended ASM1. MLVSS can vary significantly due to the influent wastewater characterization or due to seeding of activated sludge tanks. As mentioned earlier, the heterotrophic biomass concentration is a fundamental parameter for ASM1 model equations, calibration, and simulations.  $X_{BH}$  is calculated using Eq. (5). The equation parameter depends on neither HRT nor DO variability. As such, during the set of runs of HRT and DO,  $X_{BH}$  was taken as the average value calculated earlier for this specific WWTP. As per the parameters shown in Eq. (5), the value of  $X_{BH}$  depends on the MLVSS value which was reflected during the simulation runs of variable MLVSS. Fig. 6 shows the analysis of soluble COD at different sets of runs for different MLVSS concentrations and temperature ranges using fixed operational conditions of pH = 8, DO = 2.5 mg/L, HRT = 6 h, and flow rate of 2,083 m<sup>3</sup>/h. The results demonstrate that

at the same temperature, and while increasing the MLVSS concentration from 2,500 to 4,000 mg/L inside the aeration tank, the average removal efficiency of COD increases. This increase can be attributed to the increase in  $X_{BH}$  [10,11]. In addition, the results show that the average COD removal efficiency increases at different MLVSS concentrations at higher temperatures, which can be attributed to the significant increase in  $\mu_{\max(H)}$ . The average COD removal efficiency is also improved from 69% to about 84% as the measured temperature increases from 15°C to 35°C, as presented in Fig. 7. Moreover, Fig. 7 shows a convergence between simulated and measured total effluent COD concentrations at different MLVSS concentrations in g/m<sup>3</sup> and temperature ranges. As such, the plant operator can choose the required MLVSS concentration for COD removal based on the temperature of influent wastewater.

DO is an operating parameter that significantly affects COD effluent removal efficiency. As such, it is important to predict the effluent COD at different DO concentrations. Table 8 gives an overview of the removal efficiency values of the total COD concentrations under different DO concentrations at various temperature ranges using fixed operational conditions of pH = 8, MLVSS = 2,791 mg/L, HRT = 6 h, and flow rate of 2,083 m<sup>3</sup>/h. The simulation results show that the removal efficiency of the total COD increases with an increase in DO concentration, indicating the significance of the extended ASM1 [2]. COD removal efficiency increased by 15% increase when the measured



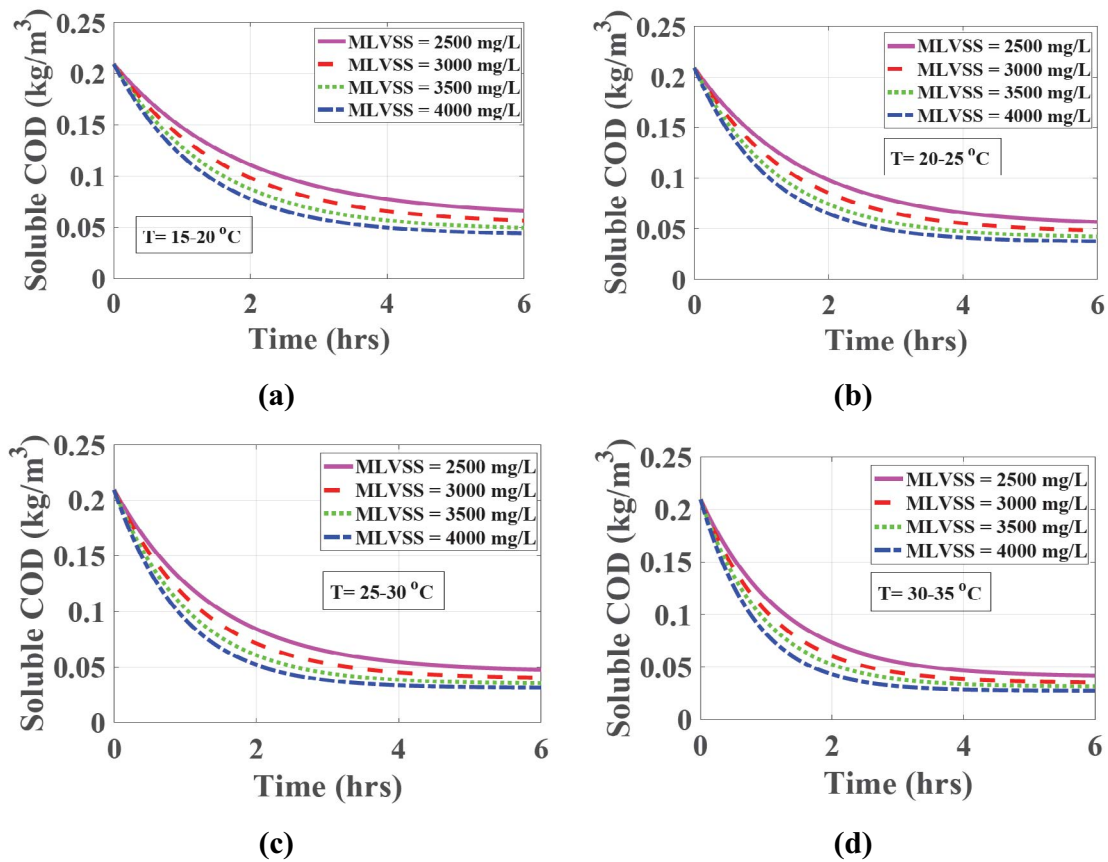


Fig. 6. MATLAB model simulations for soluble COD concentrations at different MLVSS concentrations and different temperature ranges. (a) 15°C–20°C, (b) 20°C–25°C, (c) 25°C–30°C, and (d) 30°C–35°C.

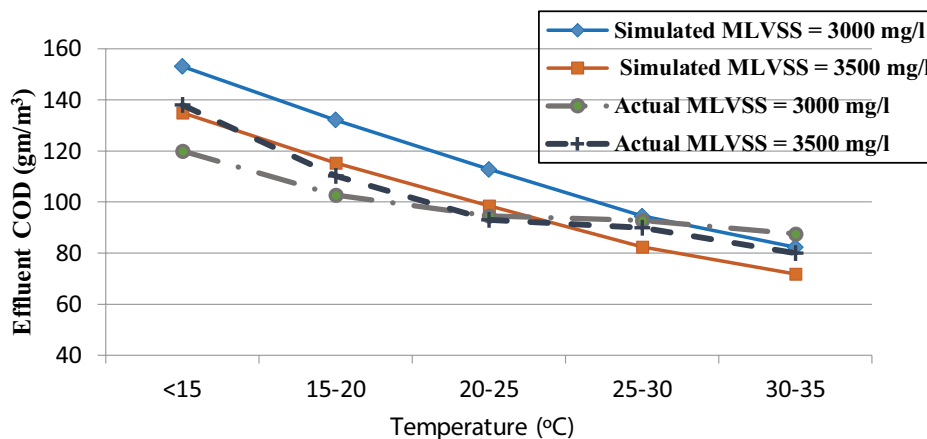


Fig. 7. Simulated and measured effluent COD concentrations for different MLVSS concentrations at different temperature ranges.

temperature increases from 15°C to 35°C. Note that the removal efficiency percentage increased by 10% as the measured temperature increased from 15°C to 25°C; however, the percentage increased by only 3% as the measured temperature increased from 25°C to 35°C.

HRT is an operating condition that significantly affects WWTP performance. Because the volume of the tanks is constant, the HRT is a function of influent discharge flowing to

the plant. Generally, most of the sewerage systems in Egypt are combined, receiving both storm and sewage discharges. During a stormwater event, the WWTP may experience diluted wastewater that can change the characteristics of the influent wastewater. In that case, HRT decreases. Conversely, HRT may increase in cases of operational complications within the factories served by the plant. A typical HRT value for the WWTP ranges from 6.14 to 6.47 h. Thus,

Table 8  
Analysis of the total COD removal efficiency for different DO values at different temperature ranges

COD removal efficiency at different DO values and temperature ranges					
DO (g/m <sup>3</sup> )	<15°C	15°C–20°C	20°C–25°C	25°C–30°C	30°C–35°C
1.0	0.589	0.639	0.69	0.737	0.771
1.5	0.623	0.672	0.719	0.763	0.794
2.0	0.643	0.69	0.734	0.777	0.81
2.5	0.654	0.7	0.743	0.785	0.813
3.0	0.662	0.707	0.75	0.79	0.817
Average	0.634	0.682	0.73	0.77	0.801

Table 9  
Analysis of the total COD removal efficiency for different HRT values at different temperature ranges

COD removal efficiency at different HRT and temperature ranges					
HRT (h)	<15°C	15°C–20°C	20°C–25°C	25°C–30°C	30°C–35°C
4.0	0.544	0.59	0.64	0.69	0.72
5.0	0.69	0.64	0.68	0.73	0.76
6.0	0.626	0.674	0.72	0.765	0.795
7.0	0.647	0.696	0.74	0.785	0.814
8.0	0.67	0.72	0.76	0.8	0.83
Average	0.635	0.664	0.708	0.754	0.784

it was decided to study the HRT under ranges that vary from 4 to 8 h utilizing the calibrated parameters. Table 9 shows the variation of the total COD removal efficiency at increasing HRTs under different temperature ranges. The results show that the average total COD removal efficiency increases when the temperature increases. At the same temperature, increasing the HRT increases the total COD removal efficiency by an average of 12% [2].

#### 4. Conclusions

An extended ASM1 was utilized to simulate a full-scale WWTP performance at various operating temperatures. Wastewater characterization was determined with a detailed sampling campaign for the WWTP influent, raw primary settled, and effluent discharge. Kinetic and stoichiometric parameters were calibrated to represent the 6th of October WWTP, located in Cairo, Egypt. After calibration and validation of the extended ASM1, several sets of runs were conducted to study the effect of temperature on total COD removal at various operating parameters: MLVSS, DO, and HRT. The results of the simulation runs were plotted vs. the temperature ranges, indicating the removal efficiency for the total COD concentration. Conversely, each parameter was studied separately within its set of runs. The runs of MLVSS showed that the value of total COD removal efficiency increased from 72% to 81% as the MLVSS concentration increases from 2,500 to 4,000 mg/L at 20°C–25°C. The simulation results also showed that the removal efficiency of total COD concentration increased by about 15% when the measured temperature increased from 15°C to 35°C at

MLVSS of 2,500 mg/L, DO of 2 mg/L, and HRT of 6 h. Note that the increase in the removal efficiency for all operating parameters was significant at temperatures ranging from 15°C to 25°C. This paper provides the operator of the WWTP with various operating scenarios that can help optimize the COD removal efficiency of the plant.

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#### Symbols

ASM1	—	Activated Sludge Model No. 1
$S_i$	—	Soluble inert COD
$S_s$	—	Soluble (readily) biodegradable COD
$X_s$	—	Particulate (slowly) biodegradable COD
$X_{B,H}$	—	Particulate heterotrophic biomass
$X_{B,A}$	—	Particulate autotrophic biomass
$X_i$	—	Particulate inert COD
$X_p$	—	Particulate bio-P heterotrophic biomass
PST	—	Primary settling tank
FST	—	Final settling tank
TSS	—	Total suspended solids, mg/L
VSS	—	Volatile suspended solids, mg/L
RAS TSS	—	Return activated sludge total suspended solids, mg/L

RAS VSS	– Return activated sludge volatile suspended solids, mg/L
MLSS	– Mixed liquor suspended solids, mg/L
MLVSS	– Mixed liquor volatile suspended solids, mg/L
TS	– Total solids, mg/L
TVS	– Total volatile solids, mg/L
TDS	– Total dissolved solids, mg/L
F/M ratio	– Food to microorganism ratio
BOD <sub>5</sub>	– Biochemical oxygen demand, mg/L
COD	– Chemical oxygen demand, mg/L
DO	– Dissolved oxygen, mg/L
SRT	– Solids retention time, day
HRT	– Hydraulic retention time, h
TKN	– Total kjeldahl nitrogen, mg/L
NO <sub>2</sub> -N	– Nitrite–nitrogen, mg/L
NO <sub>3</sub> -N	– Nitrate–nitrogen, mg/L
NH <sub>3</sub> -N	– Ammonia–nitrogen, mg/L
Total P	– Total phosphorus, mg/L
Ortho P	– Ortho phosphorus, mg/L
$C_{S(H,N)}$	– COD and NH <sub>4</sub> -N concentrations in bulk solution, kg/m <sup>3</sup>
$C_{X(H,A)}$	– Heterotrophic and autotrophic biomass concentrations in bulk solution, kg/m <sup>3</sup>
$C_{Si(H,N)}$	– Influent COD and NH <sub>4</sub> -N concentrations, kg/m <sup>3</sup>
$C_{Xi(H,A)}$	– Influent heterotrophic and autotrophic biomass concentrations, kg/m <sup>3</sup>
$C_{HS(H,N)}$	– Unionized COD and NH <sub>4</sub> -N substrate concentrations, kg/m <sup>3</sup>
$C_{O_2}$	– Dissolved oxygen concentration, kg/m <sup>3</sup>
$F_{in}$	– Influent flow to reactor, m <sup>3</sup> /h
$V$	– Volume of aeration tank, m <sup>3</sup>
$K_{d(H,A)}$	– Decay rate for heterotrophs and autotrophs, d <sup>-1</sup>
$\mu_{max(H,A)}$	– Maximum specific rate of growth for heterotrophs and autotrophs, d <sup>-1</sup>
$K_S$	– Monod half-saturation coefficient for heterotrophs, gCOD/m <sup>3</sup>
$K_{O_2(H,A)}$	– Oxygen half saturation coefficient for heterotrophs and autotrophs, g O <sub>2</sub> /m <sup>3</sup>
$K_A$	– Dissociation coefficient, dimensionless
$K_N$	– Monod half-saturation constant for autotrophs, g NH <sub>3</sub> /m <sup>3</sup>
$\mu_{max(20)}$	– Maximum specific growth rate of bacteria at 20°C, d <sup>-1</sup>
$\mu_{max(T)}$	– Maximum specific growth rate of bacteria at temperature $T$ , d <sup>-1</sup>
$\theta$	– Temperature-activity coefficient
$f_E$	– Fraction of heterotrophic active biomass that is endogenous residue
$f_{CV}$	– COD to VSS ratio of the mixed liquor organic suspended solids, mg COD/mg VSS
$f_N$	– TKN to VSS ratio of the mixed liquor organic suspended solids, mg TKN/mg VSS
$Y_H$	– Yield coefficient of heterotrophs, g COD formed/g COD oxidized
$Y_A$	– Yield coefficient of autotrophs, g COD formed/g N oxidized
OUR	– Oxygen uptake rate, mg O/L/h
$X_{BH}$	– Heterotrophic active biomass concentration due to mixed liquor, mg VSS/L

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