

Photocatalytic degradation of metronidazole (MNZ) antibiotic in aqueous media using copper oxide nanoparticles activated by H₂O₂/UV process: biodegradability and kinetic studies

Abdolmotaleb Seid-Mohammadi^a, Zeinab Ghorbanian^b, Ghorban Asgari^a,
Abdollah Dargahi^{c,d,e,*}

^a*Social Determinants of Health Research Center (SDHRC), Environmental Health Engineering, School of Public Health, Hamadan University of Medical Sciences, Hamadan, Iran, Tel. +989183729180; email: sidmohammadi@umsha.ac.ir (A. Seid-Mohammadi), Tel. +989123609313; email: asgari@umsha.ac.ir (G. Asgari)*

^b*M.Sc. of Environmental Health Engineering, School of Public Health, Hamadan University of Medical Sciences, Hamadan, Iran, Tel. +989186067993; email: zeinab.ghorbanian1003@gmail.com (Z. Ghorbanian)*

^c*Department of Environmental Health Engineering, School of Public Health, Ardabil University of Medical Sciences, Ardabil, Iran, Tel. +989141597607; email: a.dargahi29@yahoo.com (A. Dargahi)*

^d*Social Determinants of Health Research Center, School of Health, Ardabil University of Medical Sciences, Ardabil, Iran*

^e*Environmental and Occupational Health Research Center Research Center, School of Health, Ardabil University of Medical Sciences, Ardabil, Iran*

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ABSTRACT

Metronidazole (MNZ) has identified as a remarkable environmental pollutant since it is highly consumed and is resistant to biological degradation. This study was accomplished to assess the photocatalytic degradation of MNZ in aqueous media using copper oxide nanoparticles (CuO-NPs) activated by H₂O₂ in the presence of UV irradiation (UV/H₂O₂/CuO-NPs process). For this purpose, the laboratory-scale experiments were carried out in the 2.5 L batch reactor. The effect of operational parameters, that is, initial pH (3–11), the concentration of MNZ (25–100 mg/L), the concentration of H₂O₂ (10–40 mg/L), the concentration of CuO-NPs (2–8 mg/L), and reaction time (0–60 min) were also assessed. The results of this study clarified that under optimum conditions (pH = 3, H₂O₂ = 10 mg/L, [MNZ]₀ = 50 mg/L, CuO-NPs = 8 mg/L, and reaction time = 60 min), the removal efficiencies of MNZ, chemical oxygen demand (COD), and total organic carbon (TOC) using the UV/H₂O₂/CuO-NPs process were 98.36%, 73.0%, and 56.52%, respectively. According to the results, by increasing reaction time from 0 to 60 min, the AOS in the effluent was increased from 1.39 to 2.38, and the COD/TOC ratio was decreased from 1.74 to 0.8. The results related to parameters of the kinetics showed that the removal of MNZ antibiotic using the studied system conforms to the pseudo-first-order kinetics ($R^2 = 0.983$), and kinetics rate constant (k) was 0.0624 min⁻¹. This study provides UV/H₂O₂/CuO-NPs process as an innovative method to degrade the MNZ antibiotic and enhance its biodegradability. We concluded that the studied process can be used as an effective and eco-friendly method in the removal of MNZ antibiotics.

Keywords: Advanced oxidation processes (AOPs); Metronidazole antibiotic; Copper oxide nanoparticles; Hydrogen peroxide; Kinetic studies; Biodegradability

* Corresponding author.